

## Institute for Materials Science Universität Stuttgart



11<sup>th</sup> December 2023, 4:30 pm

## **Prof. Dr. Olivier Mentré** UCCS, University of Lille, France

## COLLOQUIUM MATERIALS SCIENCE Winter Semester 2023/2024

Max-Planck-Institute Werner-Köster Hall (2R4) Heisenbergstr. 3 70569 Stuttgart (Gummi Bears

served at 4:15)

## Mixed oxy-fluoride anionic sublattice : A strategy for magnetic layered perovskites

The Aurivillius materials series of formula  $(Bi_2O_2)(A_{n-1}B_nO_{3n+1})$  include perovskite slabs of variable thicknesses, n = 1, 2, 3 ... separated by Bi/O layers (where A is a large 12 coordinated cation, and B is a small 6 coordinated transition metals. They are generally limited to "high oxidation state" diamagnetic B cations such as Ti<sup>4+</sup>, Nb<sup>5+</sup>, W<sup>6+</sup>. Due to the Bi<sup>3+</sup> lone-pair effect, they mostly crystallize in polar space groups associated with attractive dielectric/ferroelectric properties. One way to access magnetically-active 3d transition metal oxide B ions is to partially substitute the O<sup>2-</sup> anionic sublattice by F<sup>-</sup> anions. Besides the limited number of reported Aurivillius (n = 1) diamagnetic oxyfluorides such as  $Bi_2Ti^{4+}O_4F_2$ ,  $Bi_2Nb^{5+}O_5F$ ,  $Bi_2V^+O_5F$ the recent synthesis of  $Bi_2Co^{2+}O_2F_4$  opens wide potentialities to combine ferrolectric and magnetic properties. We prepared  $(n = 1) \operatorname{Bi}_2 M(O,F)_6$  (M = Mn<sup>2+</sup>, Fe<sup>2+</sup>, Ni<sup>2+</sup>, V<sup>4+</sup>) as single crystals and/or polycrystalline samples, returning informative data on their accurate crystal structures, the O/F ordering in supercell, the plausible vacancies, the occurrence of various structural modulations ... etc. The segregation of antagonist O<sup>2-</sup> and F<sup>-</sup> anions in distinct sub-units of the structure has a significant impact on the bandgap and electronic properties. Electric and magnetic properties will be presented through the prism of their possible coupling into multiferroics.

Meeting-Link for TU Darmstadt: https://unistuttgart.webex.com/ unistuttgart/j.php?MTID = mcaa